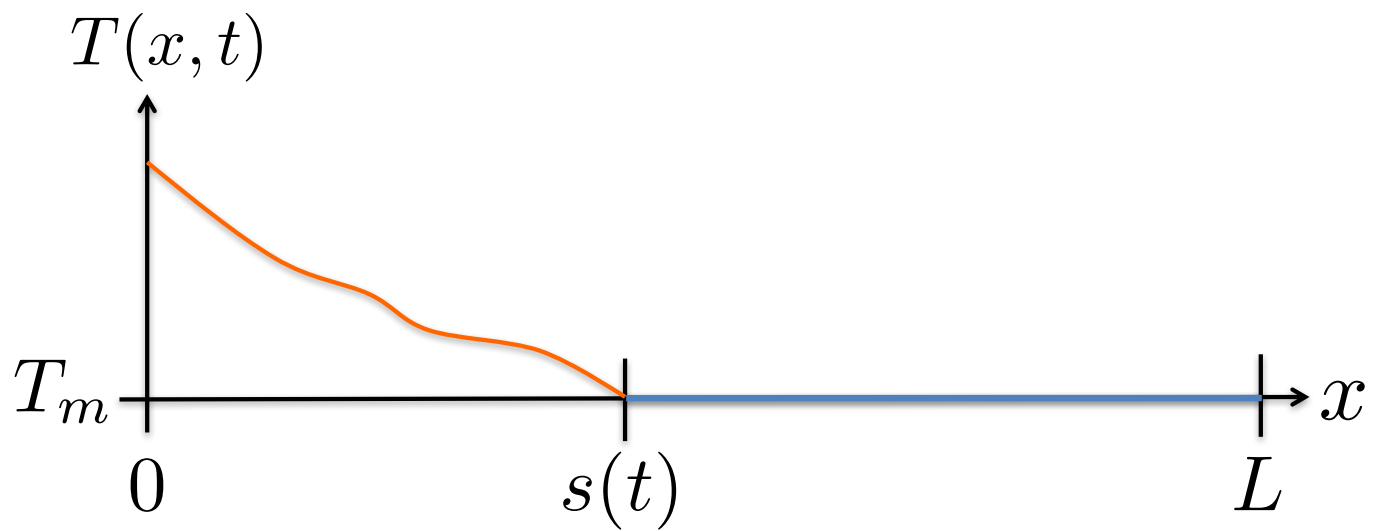
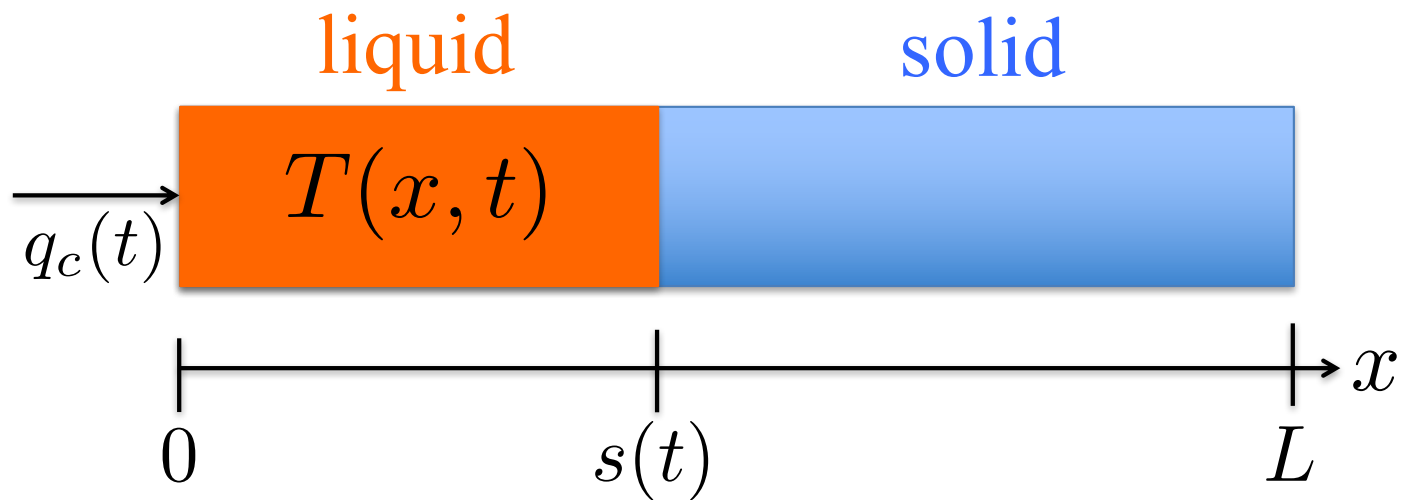


# Backstepping Control of the One-Phase Stefan Problem

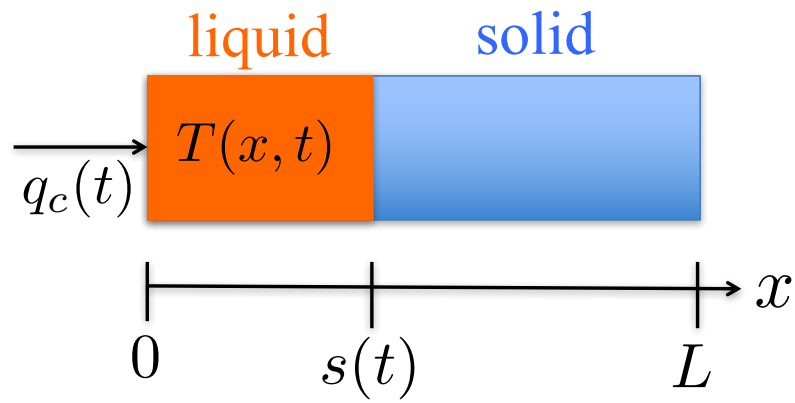
**Shumon Koga**, Mamadou Diagne, Shuxia Tang, Miroslav Krstic

University of California, San Diego

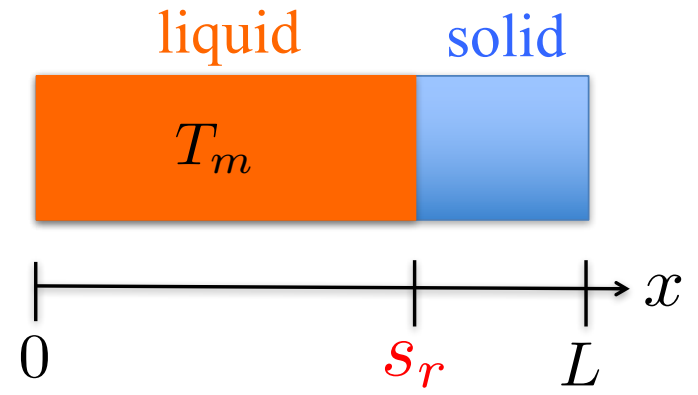
**ACC** 2016



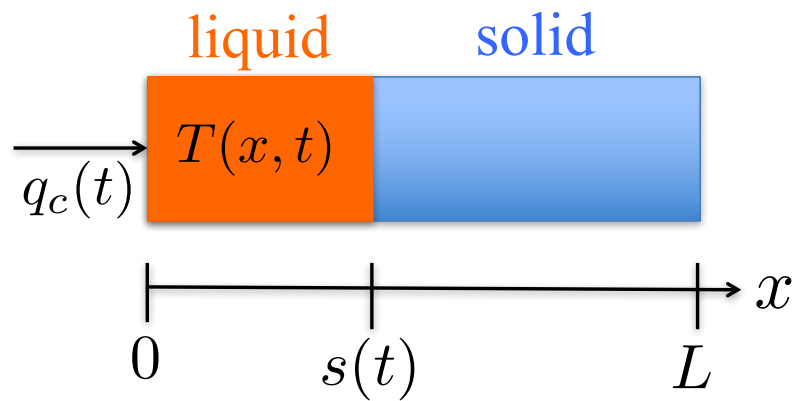
During the process



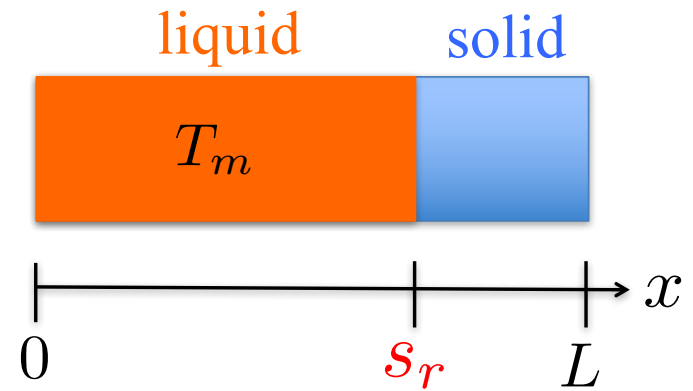
Desired state



During the process

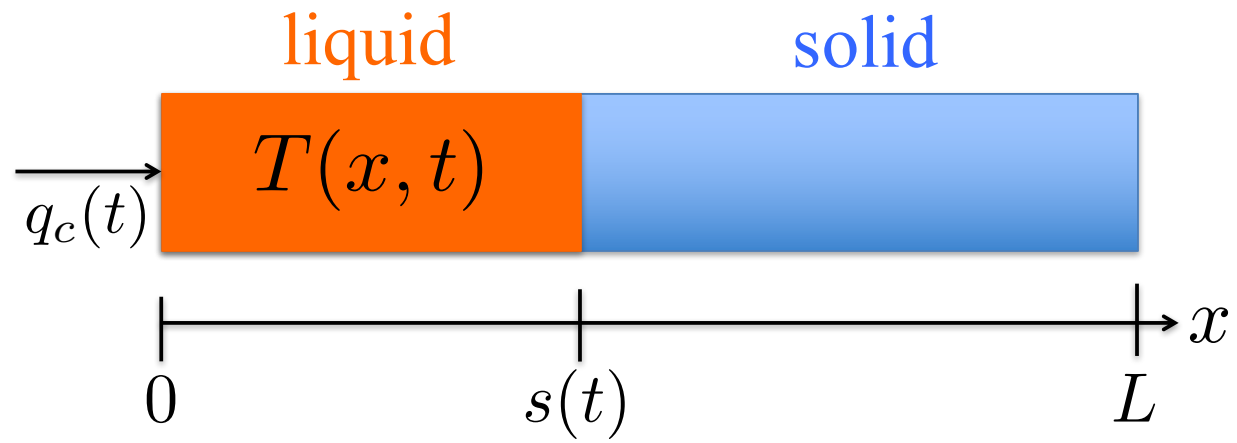


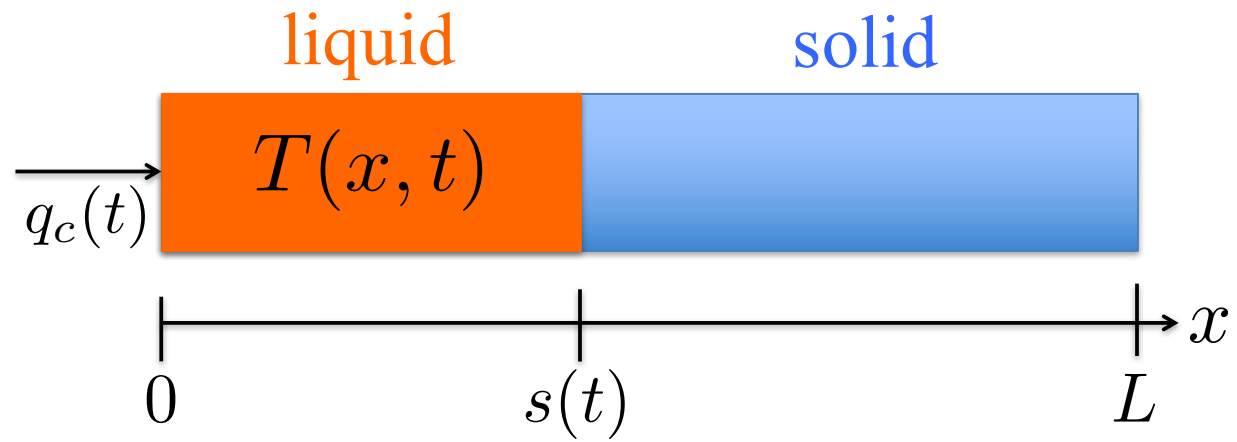
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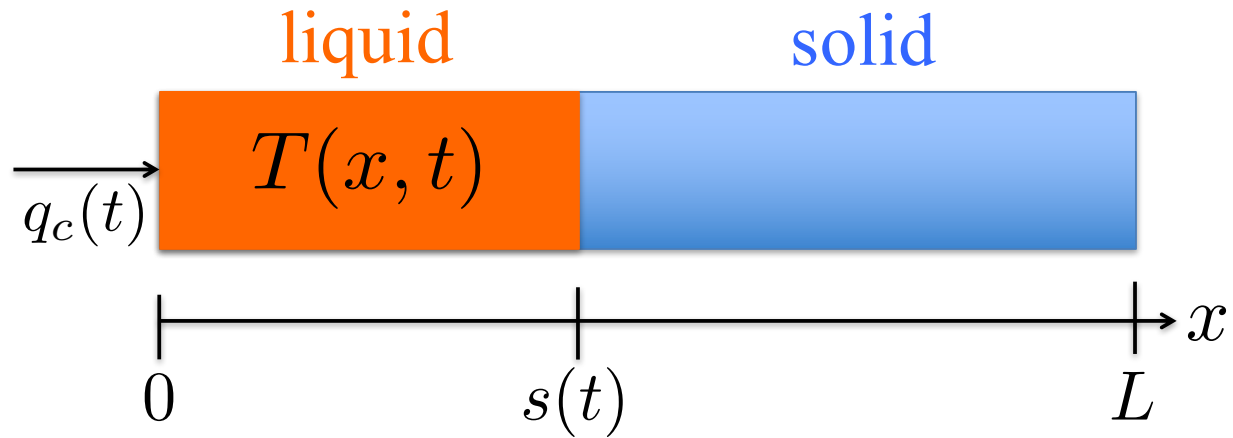
**Objective:** Design heat control  $q_c(t) > 0$  to achieve

$$s(t) \rightarrow s_r, \quad T(x, t) \rightarrow T_m, \quad \text{as } t \rightarrow \infty$$





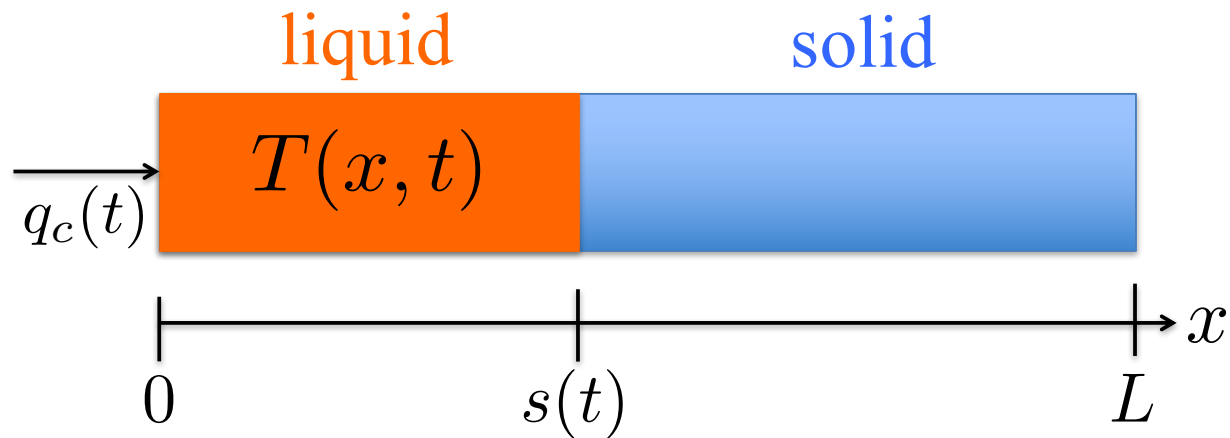
PDE  $T_t(x, t) = \alpha T_{xx}(x, t), \quad 0 < x < s(t) < L$



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$T_x(0, t) = -q_c(t)/k$

$T(s(t), t) = T_m$



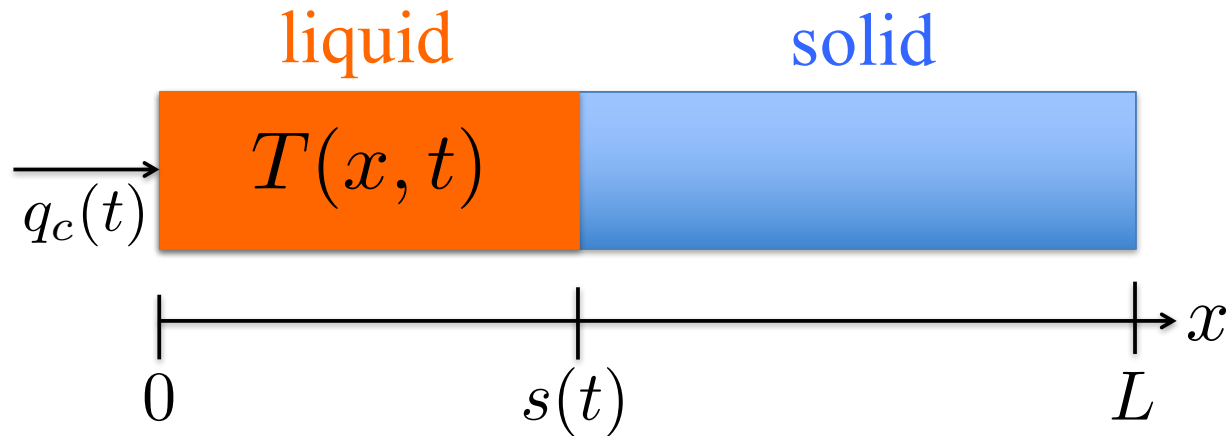
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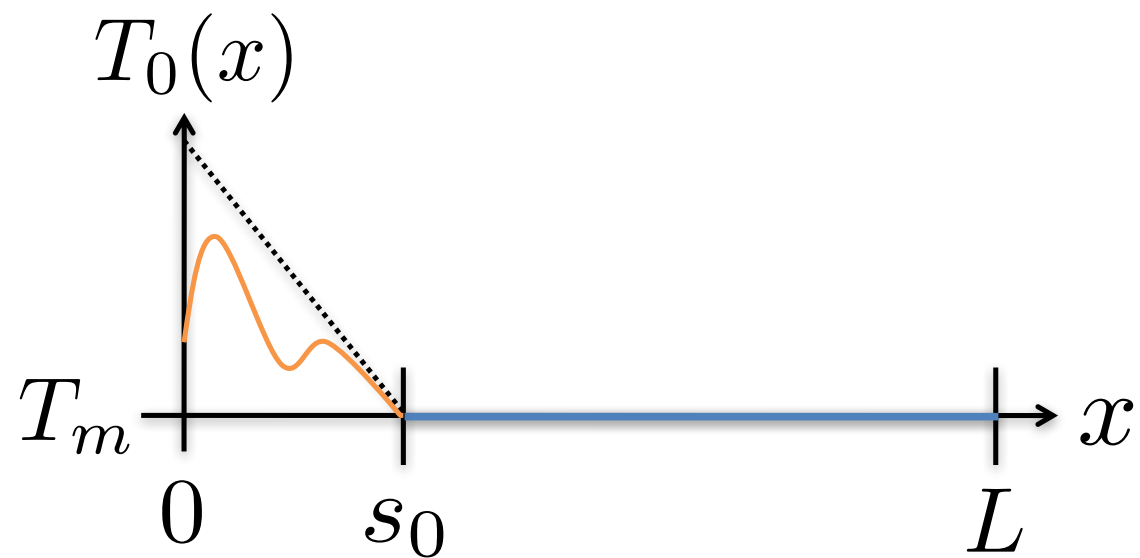
$$T(s(t), t) = T_m$$

ODE  $\dot{s}(t) = -\beta T_x(s(t), t)$

State-dependent moving boundary  $\rightarrow$  Nonlinear

**Assumption** : Initial interface position  $s_0 > 0$ , and initial temperature  $T_0(x)$  is Lipschitz ( $H := \text{Lip. const.}$ )

$$0 < T_0(x) - T_m < H(s_0 - x)$$



Model valid iff

$$T(x, t) > T_m, \quad \text{for } \forall x \in (0, s(t)), \quad \forall t > 0$$

How to guarantee this?

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How to guarantee this?

**Lemma** If  $q_c(t) > 0 \quad \forall t > 0$ , then  $\dot{s}(t) > 0 \quad \forall t > 0$  and

$$T(x, t) > T_m, \quad \forall x \in (0, s(t)), \quad \forall t > 0$$

## Energy conservation law

$$\frac{d}{dt} \left( \underbrace{\frac{1}{\alpha} \int_0^{s(t)} (T(x, t) - T_m) dx + \frac{1}{\beta} s(t)}_{\text{Internal Energy}} \right) = \underbrace{\frac{q_c(t)}{k}}_{\text{Work}} > 0$$

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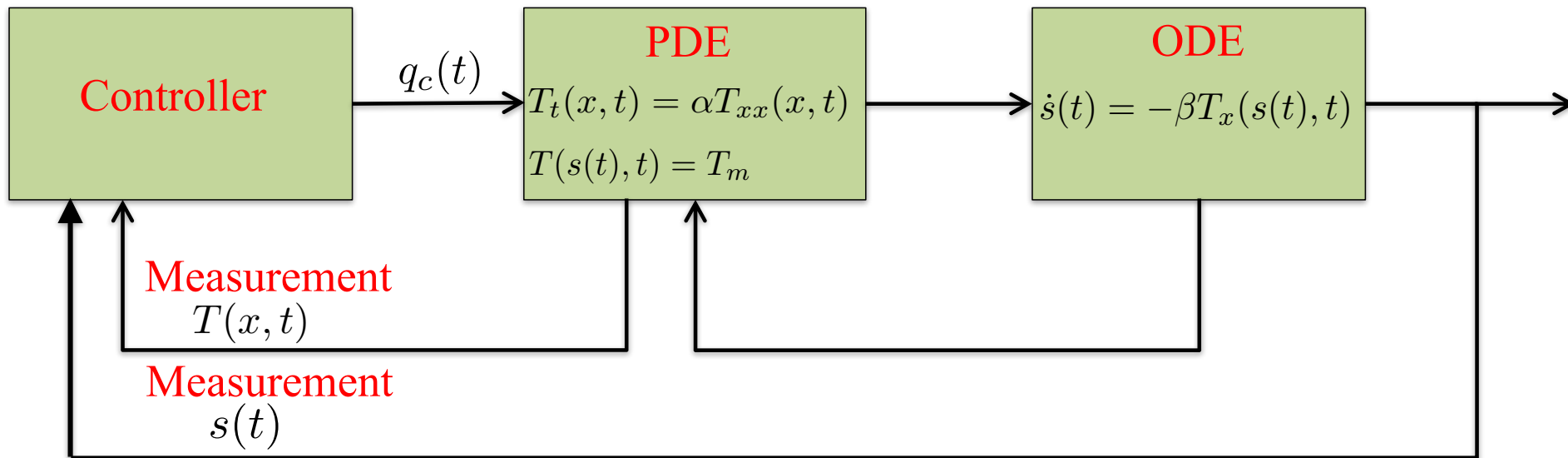
- For model to be valid (single melting interface), heat must be added.
- When heat added, internal energy grows.
- Since internal energy grows, energy corresponding to setpoint must be greater than initial energy



The following assumption **necessary** (because  $\int_0^{s_r} (T_r(x) - T_m) dx = 0$ )

**Assumption** : Setpoint  $s_r$  chosen to satisfy

$$s_r > s_0 + \frac{\beta}{\alpha} \int_0^{s_0} (T_0(x) - T_m) dx \quad =: \underline{s}_r(s_0, T_0) > s_0$$



**Theorem** The control law

$$q_c(t) = -ck \left( \frac{1}{\alpha} \int_0^{s(t)} (T(x, t) - T_m) dx + \frac{1}{\beta} (s(t) - s_r) \right)$$

where  $c > 0$ , makes the closed-loop system **globally exponentially stable** in the norm

$$\|T - T_m\|_{\mathcal{H}_1}^2 + (s - s_r)^2.$$

**Theorem** The control law

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$$\|T - T_m\|_{\mathcal{H}_1}^2 + (s - s_r)^2.$$

Note : Control law is nonlinear because of  $s(t)$  in integration limit.

## Explanation of Design

Reference errors

$$u(x, t) := T(x, t) - T_m, \quad X(t) := s(t) - s_r$$

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Backstepping transformation

$$w(x, t) = u(x, t) - \frac{c}{\alpha} \int_x^{s(t)} (x - y) u(y, t) dy + \frac{c}{\beta} (s(t) - x) X(t)$$

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Target system

$$w_t(x, t) = \alpha w_{xx}(x, t) + \frac{c}{\beta} \dot{s}(t) X(t)$$

$$w(s(t), t) = 0$$

$$w_x(0, t) = 0$$

$$\dot{X}(t) = -cX(t) - \beta w_x(s(t), t)$$

Model validity

**Proposition** Controller maintains  $q_c(t) > 0$  and  $s_0 < s(t) < s_r$ .



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Proof:

$$q_c = -ck \text{ (internal energy - setpoint internal energy)}$$

$$\dot{q}_c(t) = -cq_c(t) \quad \therefore q_c(t) = q_c(0)e^{-ct} > 0$$

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$q_c(t) > 0$  is a consequence (bonus, but not goal) of *backstepping* design!

## Lyapunov analysis

$$V_1 := \|w\|_{\mathcal{H}_1}^2 + pX^2$$

$$\begin{aligned}\dot{V}_1 &\leq -bV_1 + \dot{s}(t) \left( m_1 X(t) \|w\|_{\mathcal{L}_2} - m_2 w_x(s(t), t)^2 \right) \\ &\leq -bV_1 + a\dot{s}(t)V_1, \quad \because \dot{s}(t) > 0\end{aligned}$$

Overall Lyapunov functional

$$V := \frac{V_1}{e^{as}} = \frac{\|w(T, s)\|_{\mathcal{H}_1}^2 + p(s - s_r)^2}{e^{as}}$$

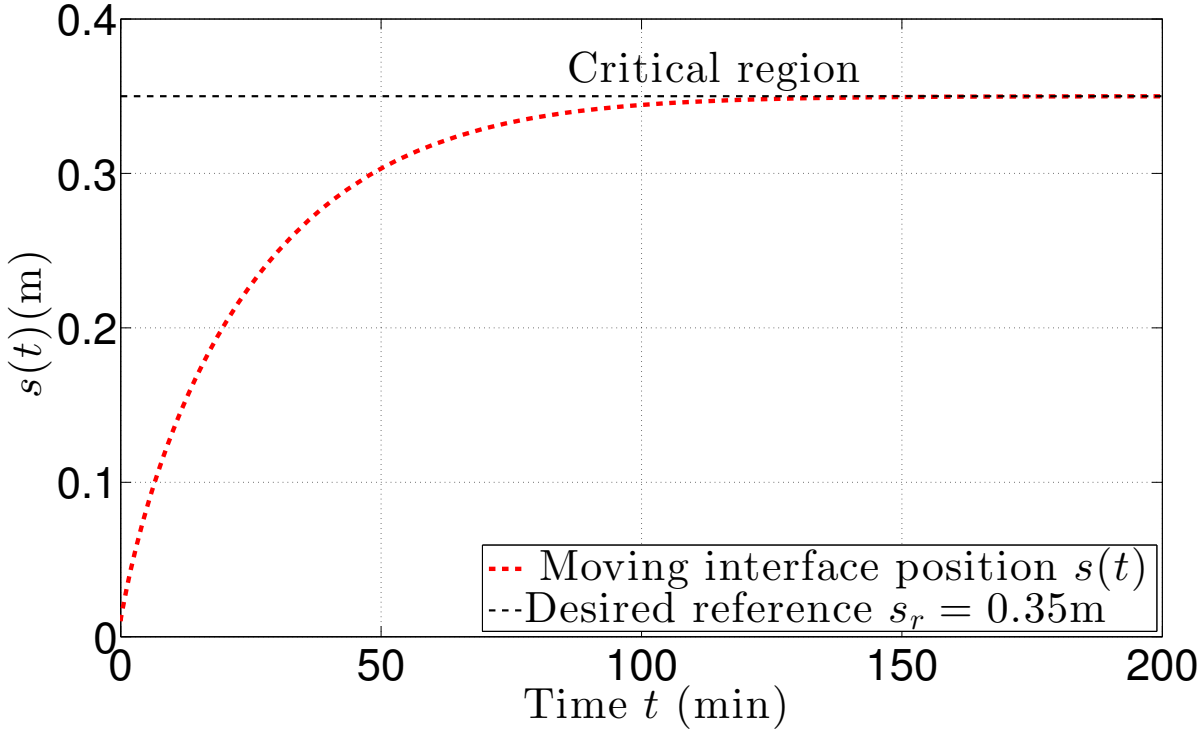
yields  $\dot{V} \leq -bV$ , which leads to

$$V_1 \leq e^{a(s(t)-s_0)} V_1(0) e^{-bt} \leq e^{a(s_r-s_0)} V_1(0) e^{-bt}$$

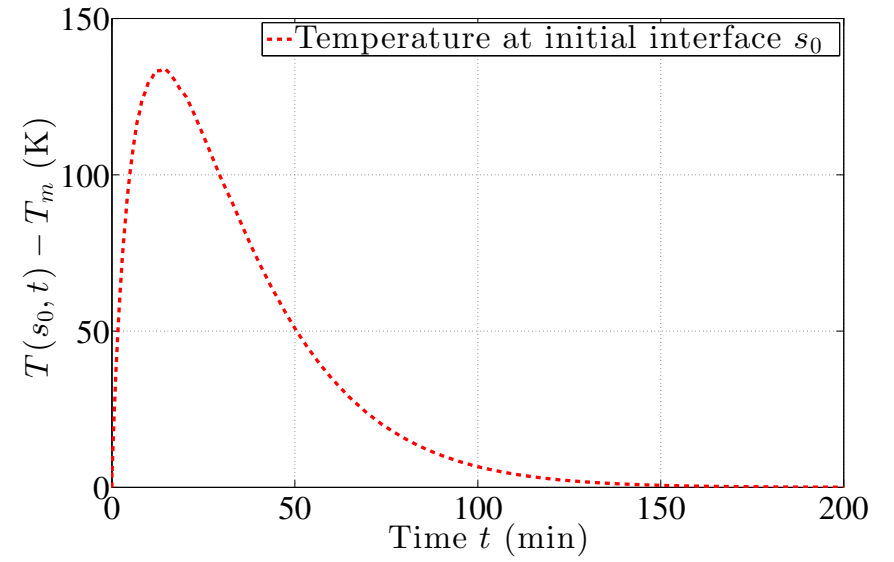
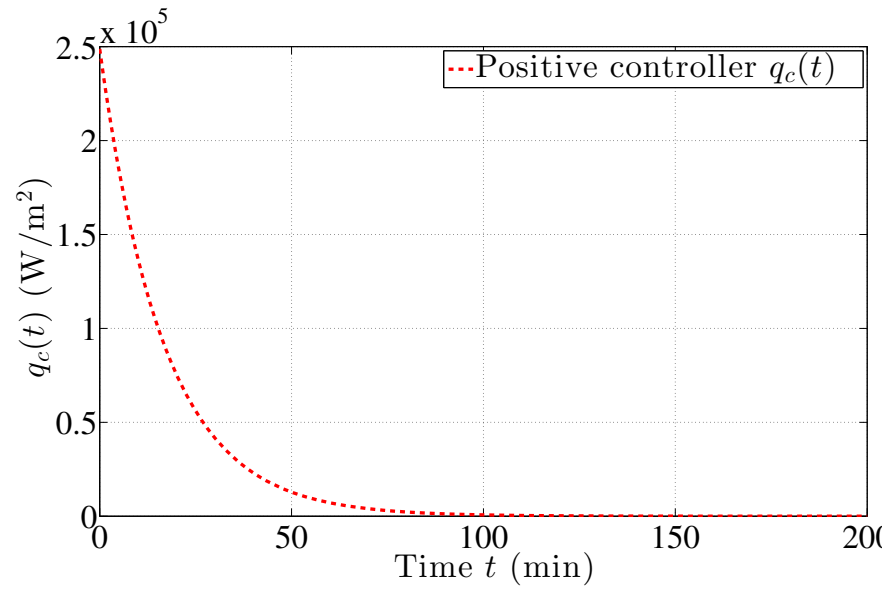
$\underbrace{\hspace{1.5cm}}_{\because s_0 < s(t) < s_r}$

# Numerical Simulation

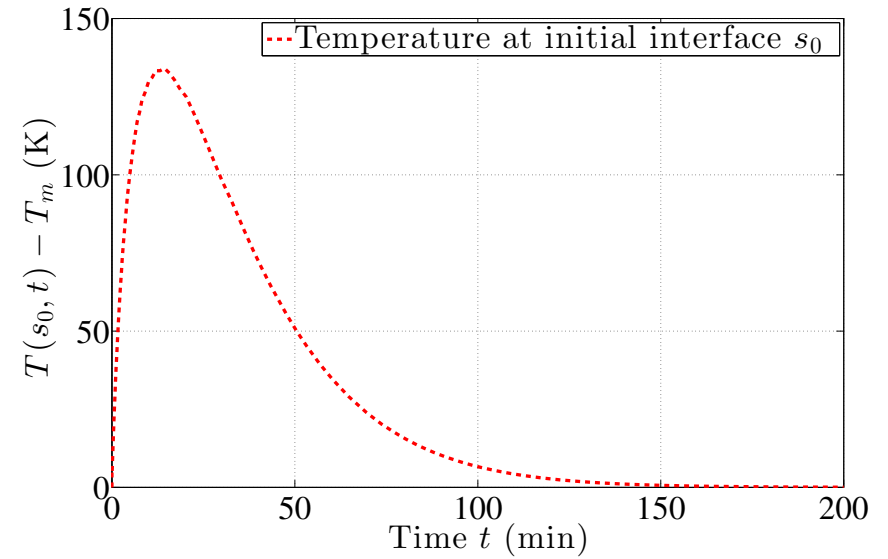
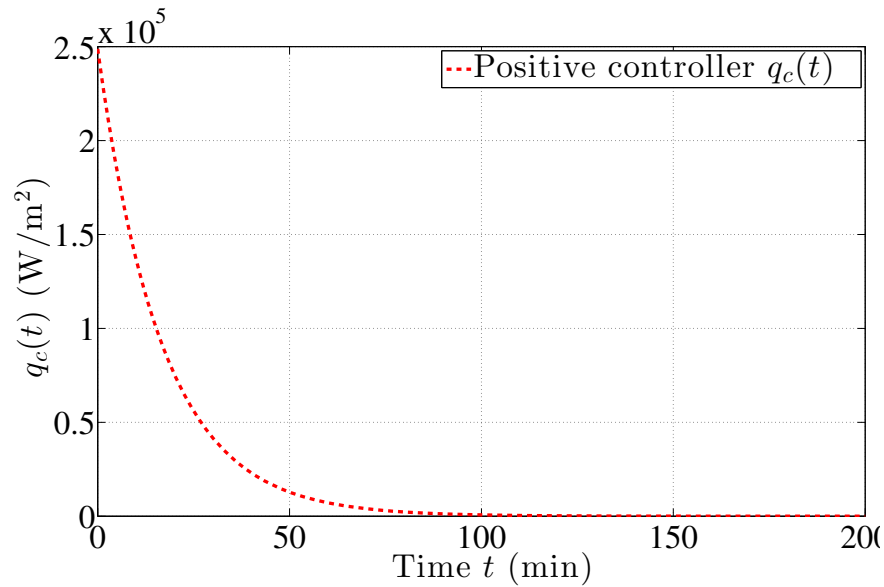
Zinc



No overshoot



Control monotonic but temperature at initial interface not monotonic.



Control monotonic but temperature at initial interface not monotonic.

Warms up from freezing/melting temperature  $T_m$ , melts the ice to  $s_r$ , and returns to freezing/melting temperature  $T_m$ .